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THE EFFECT OF CONCENTRATED SOLUTIONS OF CERTAIN MAGNESIUM SALTS ON PYOGENIC AND OTHER BACTERIA

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The feminine world for some time has known of and used a saturated solution of Epsom salt (MgSO₄) as a substitute for talcum or face powder. A small amount of the liquid is taken in the palm of the hand and rubbed gently all over the face until dry, thus leaving a soft "bloom" on the skin.

Those who were inclined to have pimples found that the use of Epsom salt had a salutary effect on the skin, the pimples drying up and disappearing. This knowledge has not been known to the writer alone. A prominent eye, ear, nose and throat specialist stated that he had gained quite a reputation as a skin specialist from suggesting this treatment after having observed its beneficial effects in several instances.

This seemingly specific action led the writer to investigate the influence of magnesium sulphate on the organism commonly found in ordinary pimples, Staphylococcus aureus.

The procedure for the phenol coefficient of MgSO₄ was carried out for Staphylococcus aureus, and also for B. typhosus with the following results:

TABLE 1
STAPHYLOCOCCUS AUREUS

Stren	igth			Time in				
Per Cent.	Dilution	2.5	5	7.5	10	12.5	15	
$\begin{array}{c}$	1:10 1:40 1:100 1:200	 _ + +	- - + +	- + +	 + +	— — + +	- + +	
MgSO ₄ 50* 25 12.5 5	1:2 1:4 1:8 1:20	+ + + +	+ + + +	+ + + +	+ + + + +	+ + + + +	+ + + + +	

^{*} Practically saturation. 50 gm. MgSO₄ were made up to 100 cc with distilled water at 25 C.

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Stren	gth		Time in Minutes											
Per Cent. Dilution		2.5	5	7.5	10	12.5	15							
(5.0	1:20	_			_									
2.5	1:40	_	-	_	_	_	_							
Phenol 1.25*	1:80			-		_	_							
111/9*	1:90	+	_	_	-	_	_							
1.0	1:100	+	+	_	_	_	_							
(0.5	1:200	+	+	+	+	+	+							
(50	1:2	+	+	+	+	+	+							
MgSO ₊ 25	1:4	+	+	+	+	+	+							
12.5	1:8	+	+	+	+	+	+							
5	1:10	+	+	+	+	+	+							

TABLE 2 B. Typhosus

These results were so unexpected and also so convincing that the experiment of growing Staphylococcus aureus in the presence of different percentages of MgSO₄ in broth was next tried.

Two hundred and fifty mm. of MgSO₄ were made up to 500 cc with nutrient broth and percentages from 50% ranging by a difference of 5%, down to 5% were prepared, filtered and autoclaved. These tubes were then each inoculated with a loopful from a broth culture of Staphylococcus aureus and incubated at 37 C. Table 3 shows the results.

1		Per Cent. of MgSO ₄ in Broth													
Age	5	10	15	20	25	30	35	40	45	50					
24 hours	+	Not ne	eessary	+	? +*	?	+	+	+	+	Macroscopic exam. Microscopic exam.				
48 hours	+	Not ne	+ cessary	+	++	++	++	++	++	++	Macroscopic exam. Growth in broth transfers made from 25-50% MgSO ₄ broth cultures				

TABLE 3
STAPHYLOCOCCUS AUREUS

The conclusion, then, is that the beneficial effect of a saturated solution of MgSO₄ on the skin is not due to any action on this particular organism.

^{*} Data for percentages 1.25% and 1 1/9% phenol were obtained from Hyg. Lab. Bul. 82, 1912, U. S. Public Health and Marine Hospital Service.

^{*} At the end of 24 hours, from the 25% on up to 50% MgSO4 cultures, transfers were made into ordinary broth to see if S. aureus was still alive. (Results in last horizontal row of the table.) It was noted that 5-25% MgSO4 broth showed typical heavy turbidity and ring formation while from 30 up to 55% a decided pellicle was formed, with less turbidity. The pellicle was heaviest on 25, 40 and 45% MgSO4 broth. Beauverie noted a similar phenomenon when growing the cholera organism in broths containing from 0.5-3% NaCl. A thicker and firmer velum was produced in the salt concentrations up to 3%.

The same concentrations of MgSO₄ in broth were inoculated with Streptococcus pyogenes, B. typhosus and B. coli—the first organism to see whether streptococcus skin infections might not be held in check by the salt, the second two organisms to ascertain whether the salt has any effect on the growth of organisms found in the intestine.

	EXI	PERIN	MENT	s v	VITH		FFER	ENT	OR-	GANI	SMS
Oncomients		I	er C	ent.	of M	[gSO					
Organisms	5	10	15	20	25	30	35	40	45	50	
Streptococcus pyogenes.		_	_	_	_	_	_	_		_*	MgSO ₄ broth Nothing visible macros- copically in plain broth
B. coli	++	++	++	++	++	?	?	?	?	?*	MgSO ₄ broth Plain broth
B. typhosus	++	++	++	++	++	?	?	?	?	?*	MgSO ₄ broth Plain broth

TABLE 4

Experiments with Different Organisms

Loop inoculations were made from the highest concentration of MgSO₄ broth that shows growth into the remaining concentrations to see whether each may become acclimatized. All of the streptococcus cultures were reinoculated.

TABLE	5
REINOCULAT	ions

Onnoniama											
Organisms	5	10	15	20	25	30	35	40	45	50	
Streptococcus pyogenes	All	nega	tive								MgSO ₄ broth Plain broth
B. coli			••	•••	*	+	::		-:-	-:-	MgSO ₄ broth Plain broth
B. typhosus		•••	•••	••	••	*		::	::		MgSO ₄ broth Plain broth

^{*} Concentrations above this were inoculated with a loopful from this concentration and incubated at 37 C. for 24 hours, reinoculated into plain broth and placed again at 37 C. for another 24 hours. Reinoculation into broth is necessary as the higher concentrations of MgSO₄ broth are too cloudy to determine growth macroscopically.

This seems to indicate that streptococcus skin infections may be inhibited if any concentration of MgSO₄ be used externally.

To determine whether MgSO₄ in a strength above saturation has a plasmolyzing effect on Staphylococcus aureus, several clean coverglasses were split with a red hot wire, flame-sterilized and placed in a

^{*} Four days after inoculation (1st 2 days at 37 C., last 2 at room temperature). At this time loop inoculations were made from each tube into broth to see if the organism can stand the various concentrations for any length of time. Broth tubes were incubated at 37 C. Results are given for each organism in the second rows above.

sterile petri dish. Then 50% MgSO₄ broth was heavily inoculated and a loopful placed on each piece of cover-glass which was put at 37 C. to dry. This took from 6-8 hours. At this time each piece was transferred to a tube of broth and incubated at 37 C.

Of the 8 tubes of broth prepared in this way 2 showed heavy growth in 24 hours, and the remaining 6 showed moderate growth as determined by turbidity. This excludes plasmolysis as a factor in the destruction of Staphylococcus aureus.

TABLE 6 MG(NO₃)₂

Organisma		Per Cent. of Mg(NO ₃) ₂												
Organisms	5	10	15	20	25	30	35	40	45	50	Hrs			
First Trial Staph. aureus	+ +	+-	- +-	=	=	=	=	=	=	=	24 48			
B. coli	++	+ -* + -	+ -* + -	‡ -* ‡ - *	+-*	+-*	+_*	+-*	+_*	+_*	24 48			
Strept. salivarius	+ -*	+-*	?	-	_	=	_	=	=	=	24 48			
B. typhosus	+* same	=	=	Ξ	=	=	_	=	=	_	24 48			
Second Trial Staph. aureus	++	++	* +	=	=	=	=	=	_		24 48			
B. coli	++	+ +*	* +	*	_	*	*	=	=	_	24 48			
B. typhosus	+	*	*	*	*	=	=	=	=	=	24 48			
Ps. pyocyanea	+ +*	*	*	*	*	* +	* +	*	=	_	24 48			

Explanations: In Tables 6, 7 and 8 + = cloudy liquid, growth present; - = clear liquid, no growth present.

The action of MgSO₄ on common pimples can hardly be explained by its physiologic action, as it has only a sedative or local anesthetic influence which is due to its depressing action on the sensory nerve endings.

Morison and Tulloch's work on the treatment of war wounds with magnesium sulphate seems to indicate that this salt has bactericidal properties. They state as follows:

^{*} Not cloudy—viscous or flocculent sediment comes up with a swirl on rotating the tube sharply.

¹ Jour. Roy. Army Med. Corps, 1916, 27, p. 375.

It possesses the desirable property of interfering with the digestive activity of pus. . . . Magnesium sulfate has not so markedly inhibitory action on phagocytosis as one would expect, and therefore, even if it be absorbed to a slight extent, it would not have a deleterious influence on the process. . . . The magnesium ion has a markedly inhibitory action on the growth of streptococci and B. coli, and a slightly inhibitory effect on the growth of B. pyocyaneus. It has, however, no easily demonstrable influence in the concentrations examined on the growth of staphylococci, nor on the diphtheroids investigated.

Similar experiments were carried out with other soluble magnesium salts Mg(NO₃)₂, MgCl₂ as well as MgSO₄, in order to determine whether the Mg ion itself is responsible for the action observed or whether this particular combination of ions is necessary. Tables 6, 7 and 8 show that the latter hypothesis is probably correct:

TABLE 7 MgCl₂

Organisms	Per Cent. of MgCl2												
Organisms	51/3	10%	161/3	211/3	26 %	32	371/3	42%	48	53 1/ 3	58 %	66%	
Staphylococcus aureus	++	++	++	++	+*	+* +	+ *	=	=	=	=	=	24 hours 48 hours
B. coli	++	+*	=	=	=	=	=	=	=	=	Ξ	=	24 hours 48 hours
B. typhosus	++	=	=	=		=	=	=	=	=	=	=	24 hours 48 hours
Ps. pyocyanea	++	=	=	_	=	=	=	=	=	Ξ	=	=	24 hours 48 hours

TABLE 8
MgSO₄

Organisms	Per Cent. of MgSO ₄											
Organisms	5	10	15	20	25	30	35	40	45	50		
Staphylococcus aureus	++	+++	++	+++	++	++	+	+++	+++	++	24 hours 48 hours	
B. coli	++	++	++	++	++	++	+	+		=	24 hours 48 hours	
Ps. pyocyanea	++	+++	++	++	++	++	++	+++	++	=	24 hours 48 hours	
B. typhosus	+	+	+++	+	++	*	=	=	=	=	24 hours 48 hours	

The results of Morison and Tulloch with regard to S. pyogenes were proven in the use of MgSO₄, but not in case of B. coli; MgCl₂ was the only magnesium salt used that had an inhibitory action. In

fact, this latter salt had the most marked inhibitory action on the growth of all organisms used in the experiment. This suggests that a substitution of MgCl₂ for MgSO₄ might have the advantage under certain conditions.

Further study of the specific action of concentrated solutions of MgSO₄ and other magnesium salts on the infected skin or in wounds may present interesting if not valuable information.